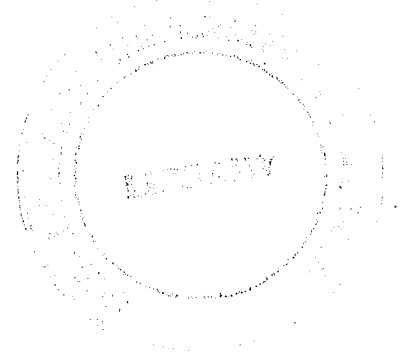


Uva Wellassa University, Sri Lanka  
Btech. Science and Technology  
End Semester Examination- Semester 1  
December -2008



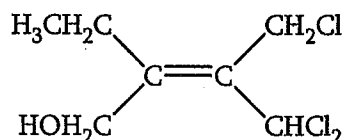
SCT 232-1 Organic Chemistry

Answer four (04) questions only

Time: One (01) hour

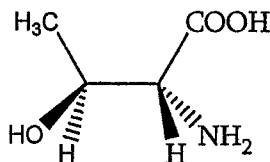
1. a. Looking along C<sub>2</sub>-C<sub>3</sub> bond of butane draw the Newman projections for the rotations from 60° and construct the energy diagram. Indicate the followings in the energy diagram:
- Most stable conformation
  - Least stable conformation
- (15 marks)
- b. Draw *trans*-1,2-dimethylcyclohexane in its most stable chair conformation.
- (4 marks)
- c. Draw structure corresponding to the following:
- 3,5-Dinitrobenzaldehyde
  - N-Methylcyclohexylamine
- (6 marks)

2. a. Assign E or Z configuration for the following compound.



(5 marks)

- b. Draw the Fisher projection corresponding to the following structure and assign R or S configuration for the chiral center(s).

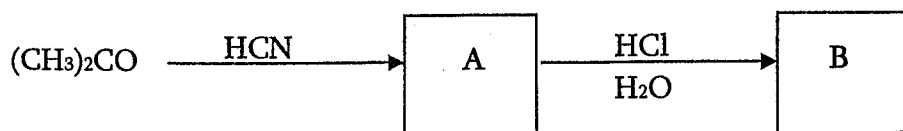


(10 marks)

- c. A 1.20 g sample of cocaine (an optically active compound) was dissolved in 7.50 mL of chloroform and placed in a sample tube with a path length of 5.00 cm. The observed rotation at the sodium D line at 25 °C was -1.3°. Calculate the specific rotation for cocaine and express it in standard notation.

(10 marks)

3. a. Classify the following reactions as addition, elimination, substitution and rearrangement.
- $\text{CH}_3\text{Br} + \text{KOH} \rightarrow \text{CH}_3\text{OH} + \text{KBr}$
  - $\text{CH}_3\text{CH}_2\text{OH} \rightarrow \text{CH}_2=\text{CH}_2 + \text{H}_2\text{O}$
  - $\text{CH}_2=\text{CH}_2 + \text{H}_2 \rightarrow \text{CH}_3\text{CH}_3$
- (6 marks)
- b. Arrange the following in order of decreasing nucleophilicity.  
 $\text{CH}_3\text{O}^-$ ,  $\text{H}_2\text{O}$ ,  $\text{CH}_3\text{OH}$ ,  $\text{HO}^-$
- (4 marks)
- c. Give the mechanism for bromination of benzene with  $\text{FeBr}_3/\text{Br}_2$
- (9 marks)
- d. What are the products expected from the bromination of
- $\text{C}_6\text{H}_5\text{NH}_2$
  - $\text{C}_6\text{H}_5\text{NO}_2$
- (6 marks)
4. a. Describe the effect of the following variables on both  $\text{S}_{\text{N}}2$ ,  $\text{S}_{\text{N}}1$  reactions:
- Substrate structure
  - Leaving group
- (10 marks)
- b. "Optically active starting materials give racemization of stereochemistry at the reaction centre by  $\text{S}_{\text{N}}1$  reactions but  $\text{S}_{\text{N}}2$  reactions give inversion of stereochemistry at the reaction centre". Explain. Use illustrations where necessary.
- (15 marks)
5. a. Draw a fully labeled reaction energy diagram for a two step reaction whose second step is faster than it's first step.
- (7 marks)
- b. Give the mechanism for the aldol reaction of  $\text{CH}_3\text{CHO}$  in the presence of  $\text{NaOH}$ .
- (10 marks)
- c. Give the structures of compounds A and B



(8 marks)